

# δ - TRIP Steel

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## Introduction

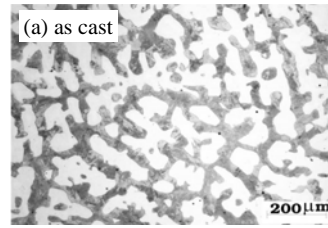
An attempt has recently been made to produce TRIP-assisted steel which contains a large fraction of δ-ferrite dendrites locked into the microstructure during the solidification process. This δ-ferrite persists in the microstructure during subsequent heat treatment, such that the steel can never become fully austenitic at any temperature. The steel is then heated into the two phase δ + γ phase field and the resulting austenite then transformed isothermally into a mixture of bainite and carbon-enriched retained austenite. This retained austenite gives the steel its improved properties through transformation-induced plasticity. This concept contrasts with normal TRIP-assisted steels in which the steel can be fully austenitised. It has been discovered in further experiments that the alloy design procedure which ensures the presence of stable δ-ferrite in the microstructure must take into account that the solid-state transformation that takes place in cooling from the solidus temperature does not occur with the equilibrium partitioning of solutes. As a consequence, the amount of δ-ferrite obtained can be far less than expected from equilibrium calculations.

## Alloy design

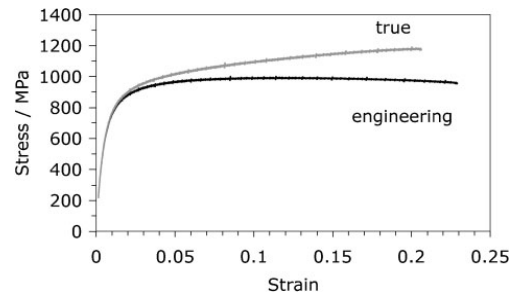
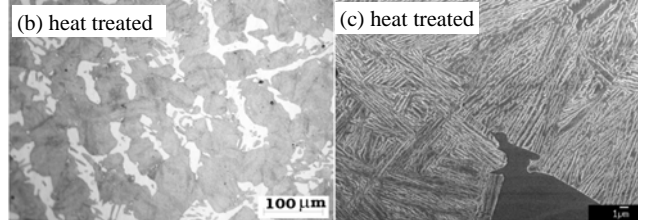
Two alloys were designed based on the phase diagram calculated by MTDATA with TCFE v1.21 database.

Chemical compositions of two designed alloys

	C	Si	Mn	Al	Cu	P
DT2_design	0.4	0.25	2.0	2.1	0.5	0.02
DT3_design	0.4	0.25	2.0	2.5	0.5	0.02



Optical micrograph (a), (b) and FEGSEM image (c), of δ-TRIP steel



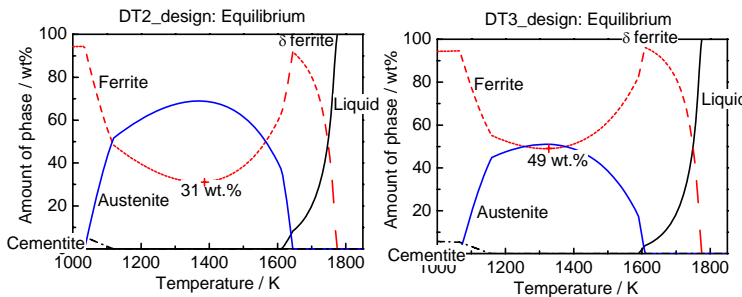
Stress-strain curve of heat treated δ-TRIP steel at room temperature (S. Chatterjee and H. K. D. H. Bhadeshia, 2007)

## Experimental

Two 34kg ingots were obtained from POSCO using a steel mold in an electric vacuum furnace with a approximate cooling rate 20°C/s, whose dimensions were 170mm in length, 100mm width and 230 mm in height, whose chemical compositions are shown the following table.

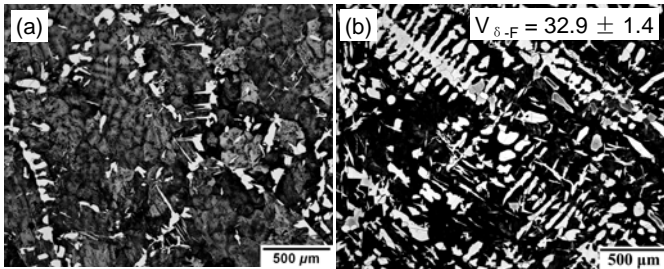
Chemical compositions of two alloys produced in POSCO

	C	Si	Mn	Al	Cu	P
DT2_P	0.361	0.255	2.02	2.13	0.485	0.02
DT3_P	0.368	0.233	1.985	2.493	0.491	0.02



Equilibrium phase diagrams for the two designed steels

## Results and Discussions



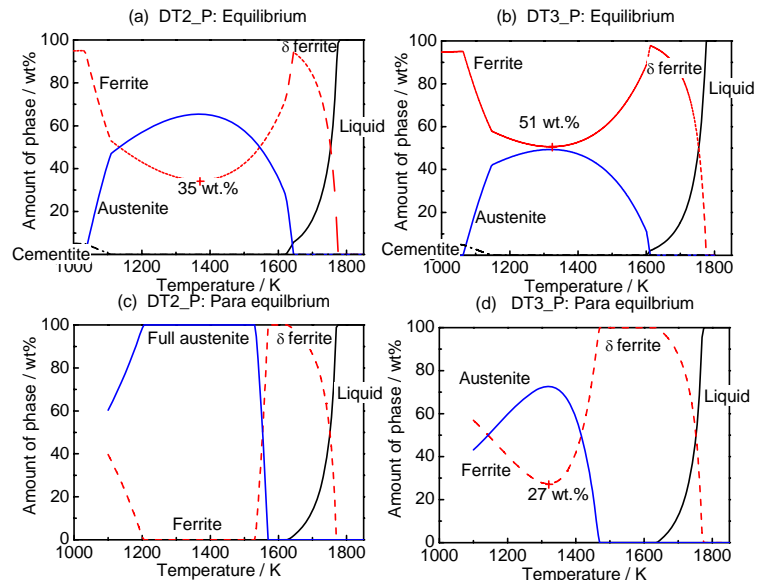
Optical micrograph of (a) DT2\_P and (b) DT3\_P in as-cast condition

## Conclusions

Aluminium is the most significant element to increase the fraction of δ-ferrite in the as cast steel. It containing 2.5 wt.% of Al fulfilled 32.9±1.4 vol.% of δ-ferrite at the as cast condition at the approximate industrial continuous cooling rate 20°C/s. The mechanism that the volume fraction of δ-ferrite is between the equilibrium and para-equilibrium phase diagrams will be discussed in the other paper.

## Acknowledgement

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Equilibrium (a), (b) and para-equilibrium (c), (d) phase diagrams for the two steels received from POSCO