



# The magnetic-field-induced precipitation behaviors of alloy carbides

T.P. Hou, Y. Li and K.M. Wu

Wuhan University of Science and Technology

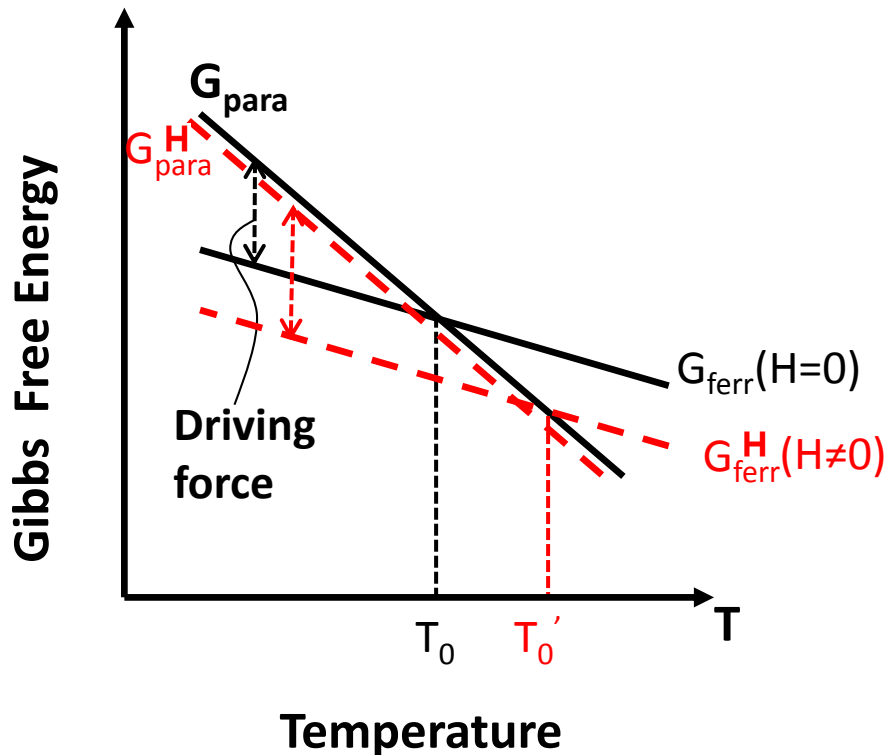
E-mail: [wukaiming@wust.edu.cn](mailto:wukaiming@wust.edu.cn)

## Acknowledgements

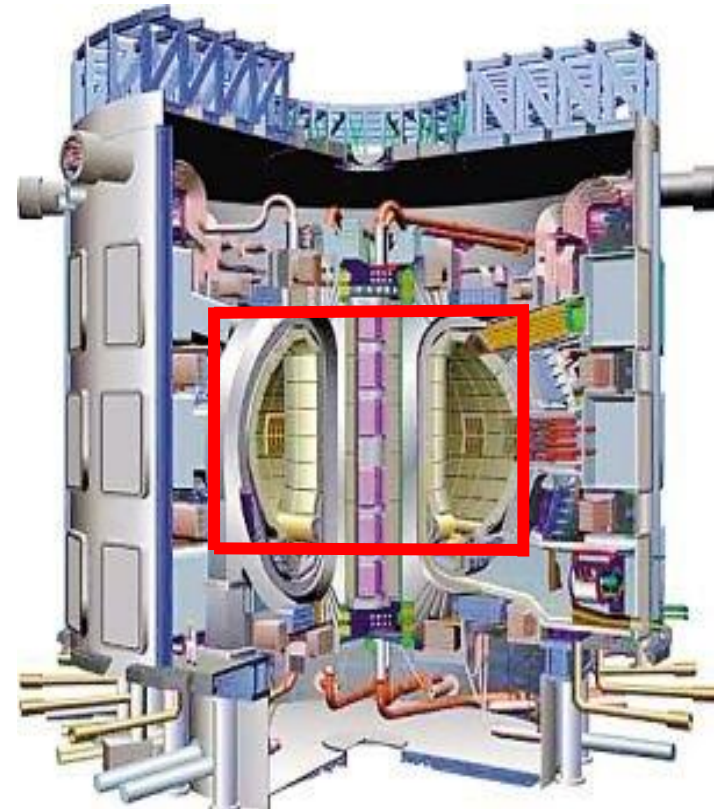
**Professor M. Enomoto, Ibaraki University, Japan, for providing alloy specimens  
State Ministry of Education (Grant No. NCET-05-0680)**

# Introduction

3-5 T, 400-550 °C



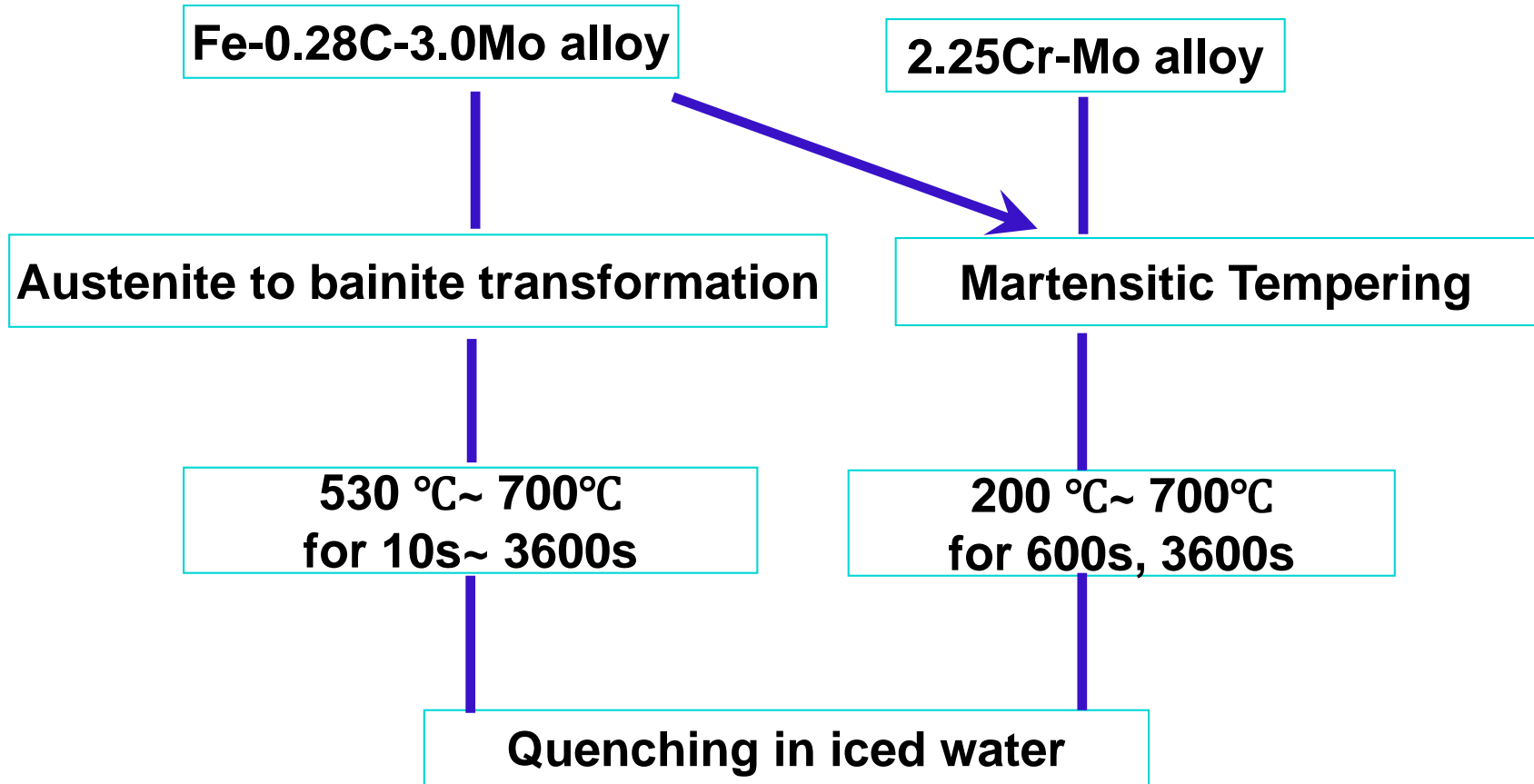
**Influence of magnetic field on phase transformation**



**Schematic diagram of the principle component in Tokamak in reduced activation steel**

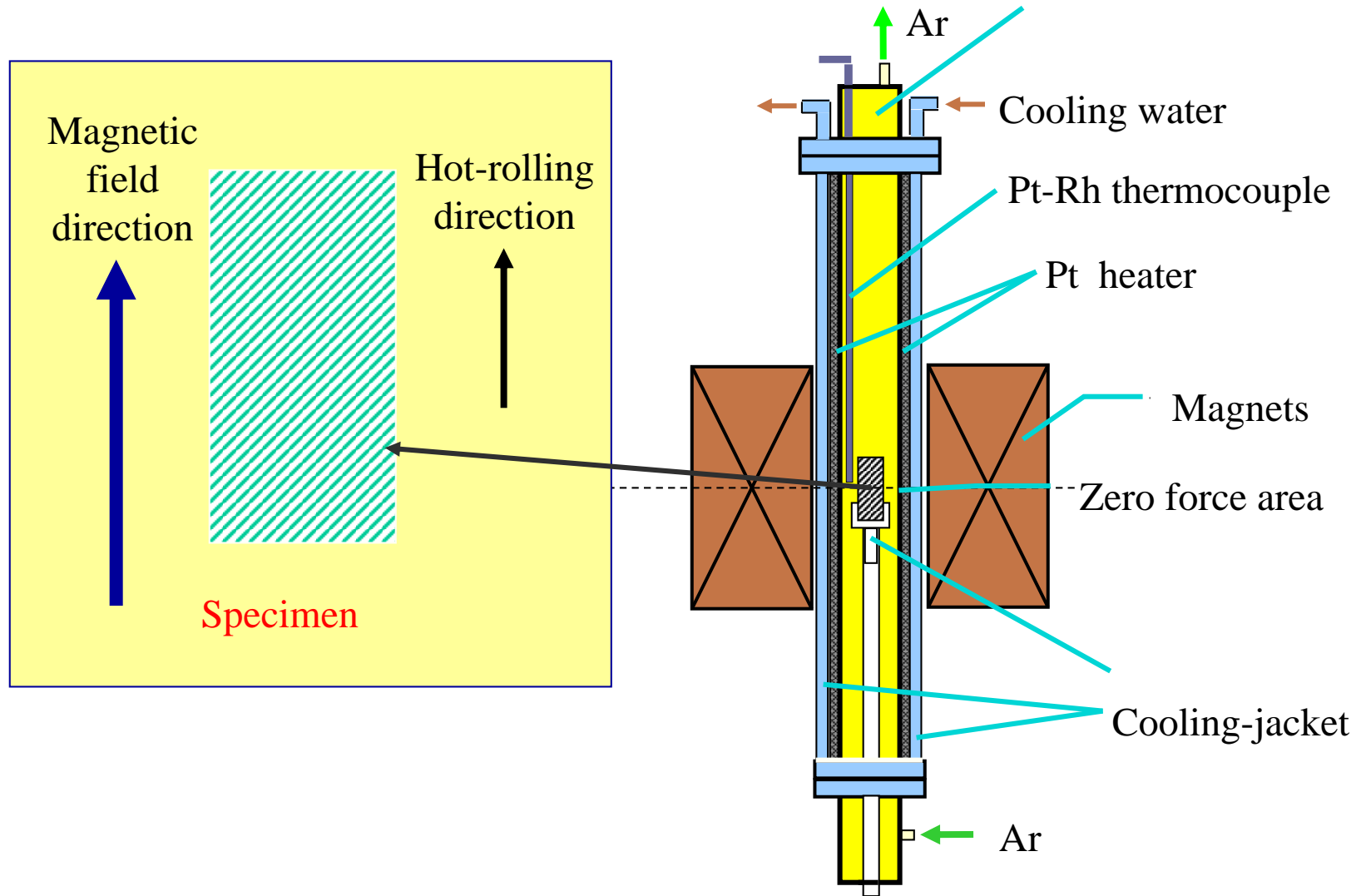
Nuclear fusion reactor Reece, R. J. 1988

# Magnetic (12T) Heat Treatment of Alloys

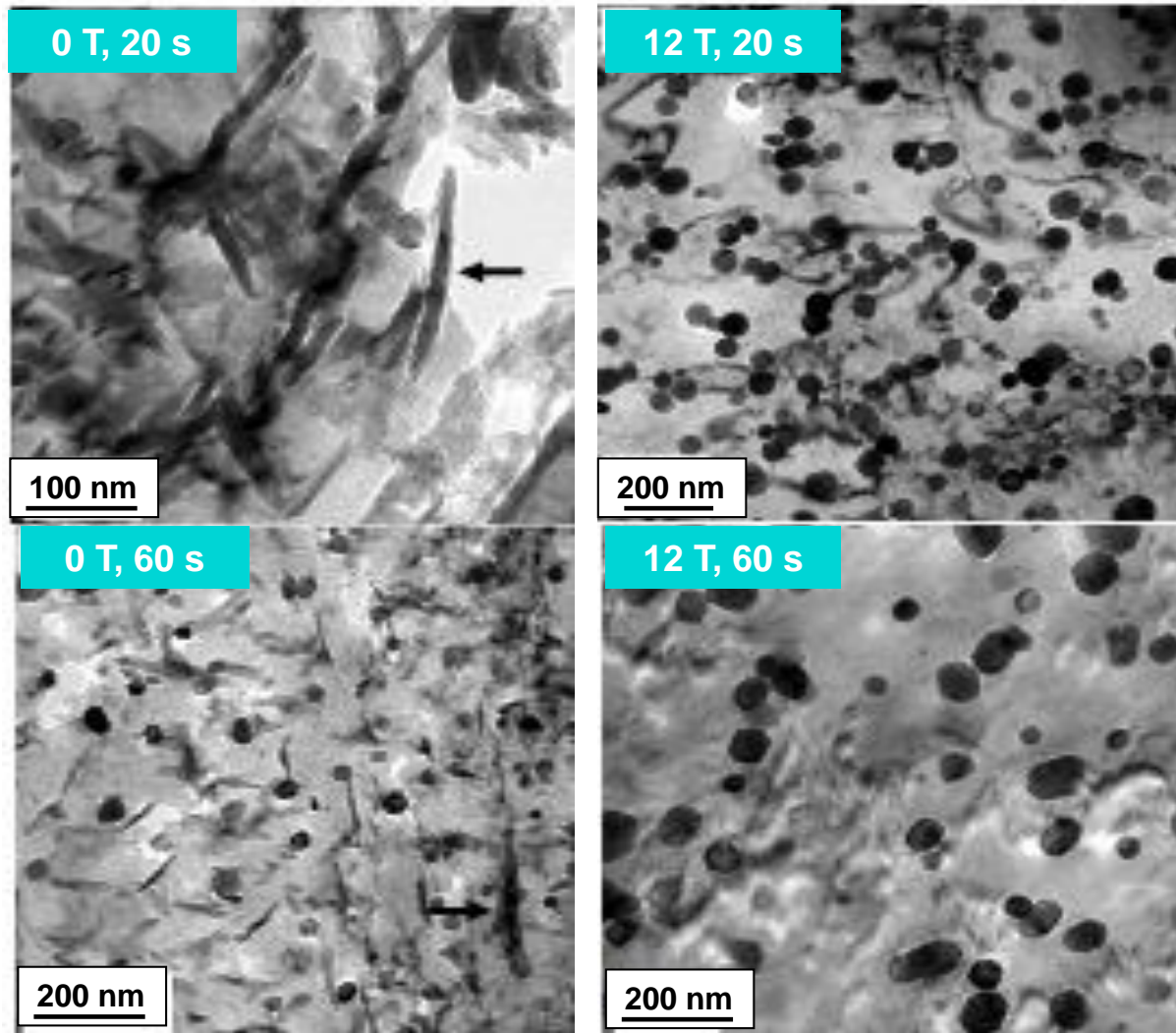


Two alloys, three heat treatments

# Magnetic Heat Treatment Equipment



# Steel I : Fe-C-Mo, $\gamma \rightarrow \alpha$ at 530°C



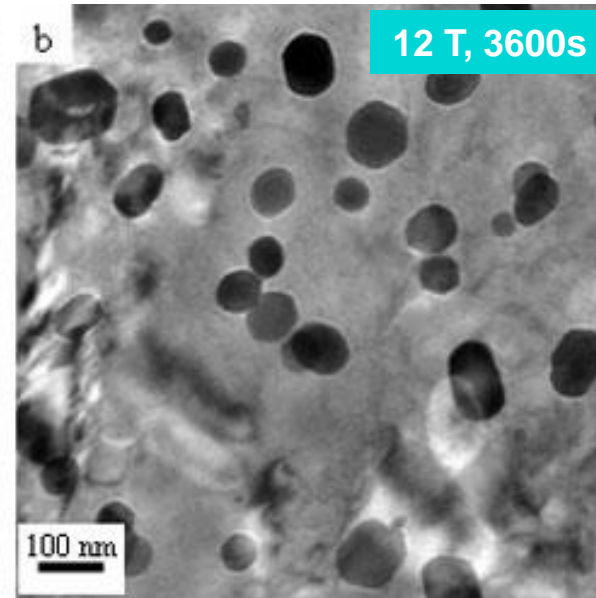
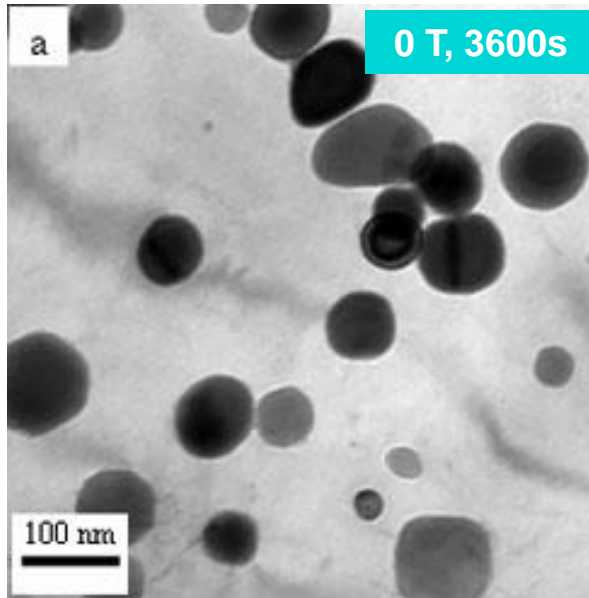
High magnetic field promotes the precipitation of  $M_6C$

# Steel I : Fe-C-Mo, $\gamma \rightarrow \alpha$ at 530°C

0 T			12 T	
Time	1000 s	3600 s	600 s	3600 s
Type	$M_2C$ , $M_3C$	$M_6C$	$M_6C$	$M_6C$

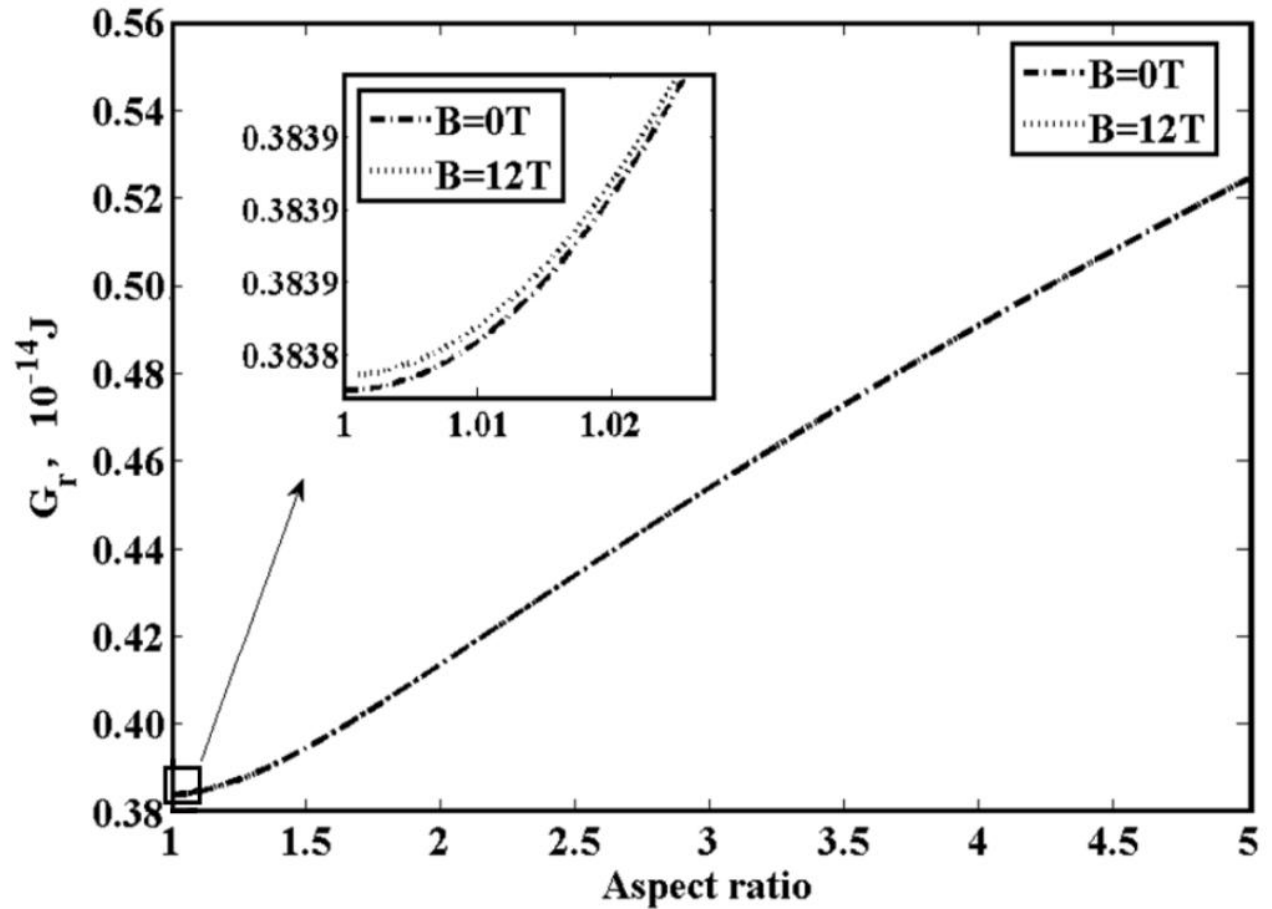
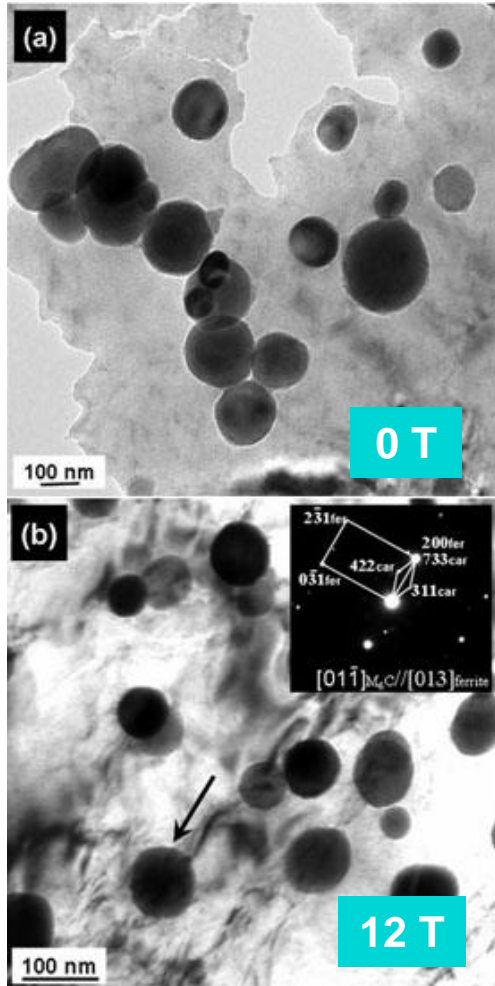


Magnetic field promotes the  $M_6C$  precipitation



Magnetic field has no influence on the morphology of  $M_6C$

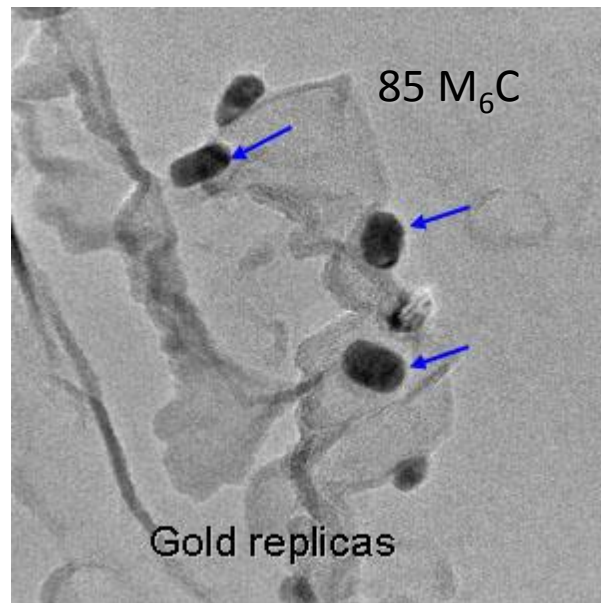
# Steel I : Fe-C-Mo, $\gamma \rightarrow \alpha$ transformation



No visible influence on alloy carbide morphology

# Steel I : Fe-C-Mo, $\gamma \rightarrow \alpha$ transformation

Temp.	0 T				12 T			
	Fe (wt.%)		Mo (wt.%)		Fe (wt.%)		Mo (wt.%)	
	Average	$\sigma$	Average	$\sigma$	Average	$\sigma$	Average	$\sigma$
530°C	38.69	0.98	61.30	0.98	67.16	4.48	32.78	4.48



EDS analysis

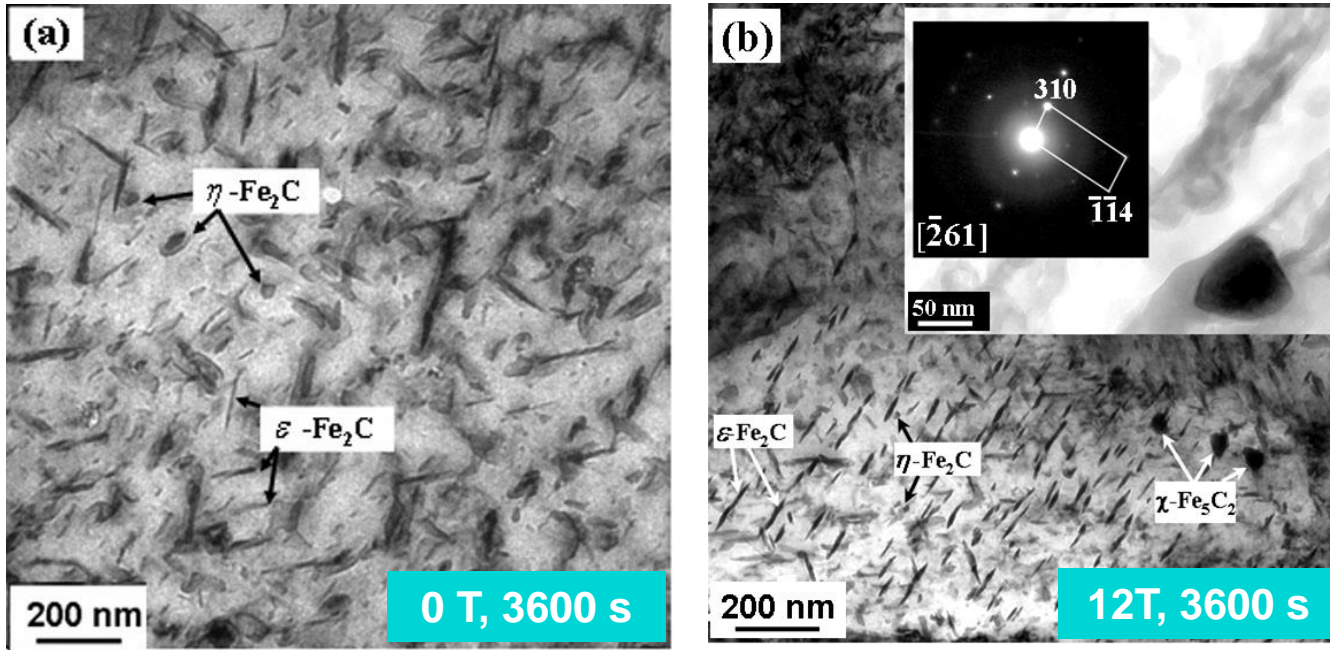
Higher than the content without the magnetic field

0 T, Fe (wt.%)	
38-45 (700°C~900°C)	Ref. [Sato T, 1962]
36.4-38.6 (727°C)	Ref.[Uhrenius B,1975]
58.15 (874°C)	Ref. [Woodyatt LR,1979]

Magnetic field increases the Fe concentration in  $M_6C$



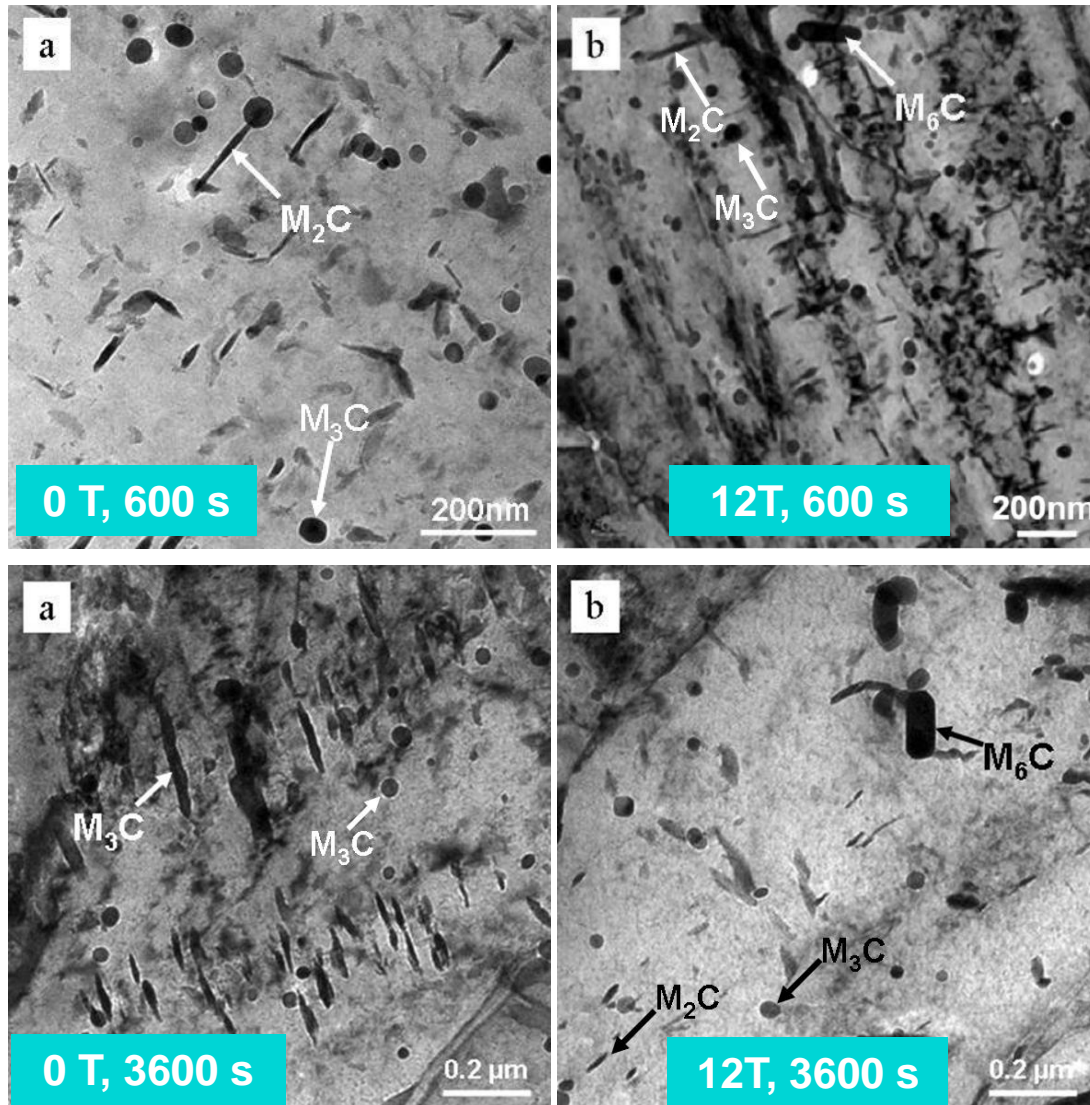
# Steel I : Fe-C-Mo, Tempered at 200 °C



	0 T		12 T	
Time	600 s	3600 s	600 s	3600 s
Carbide types	ε-Fe <sub>2</sub> C η-Fe <sub>2</sub> C	ε-Fe <sub>2</sub> C η-Fe <sub>2</sub> C	ε-Fe <sub>2</sub> C η-Fe <sub>2</sub> C	ε-Fe <sub>2</sub> C η-Fe <sub>2</sub> C <b>χ-Fe<sub>5</sub>C<sub>2</sub></b>

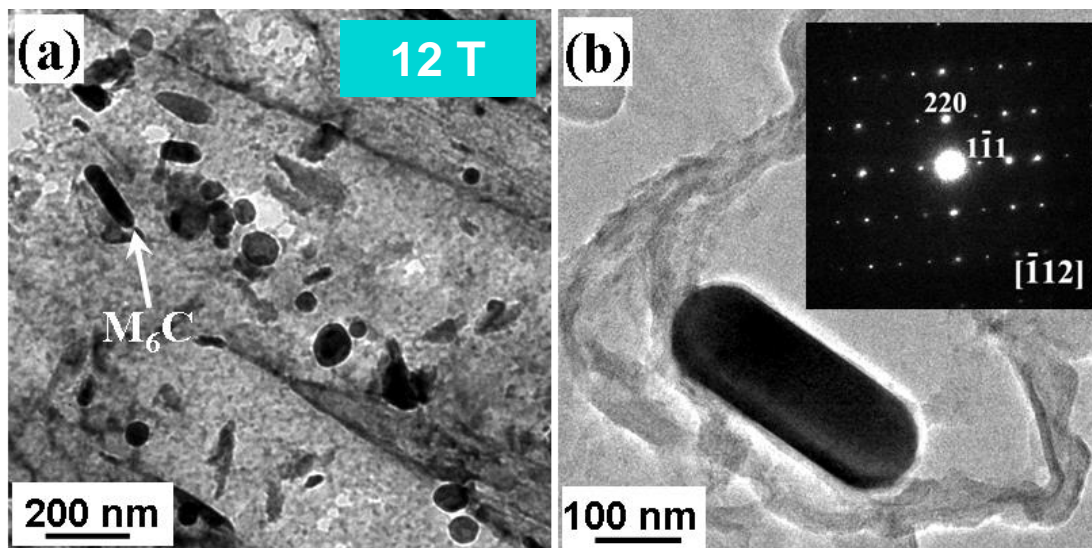
High magnetic field promotes the precipitation of  $\chi\text{-Fe}_5\text{C}_2$

# Steel I : Fe-C-Mo, Tempered at 530 °C



High magnetic field promotes the precipitation of  $M_6C$

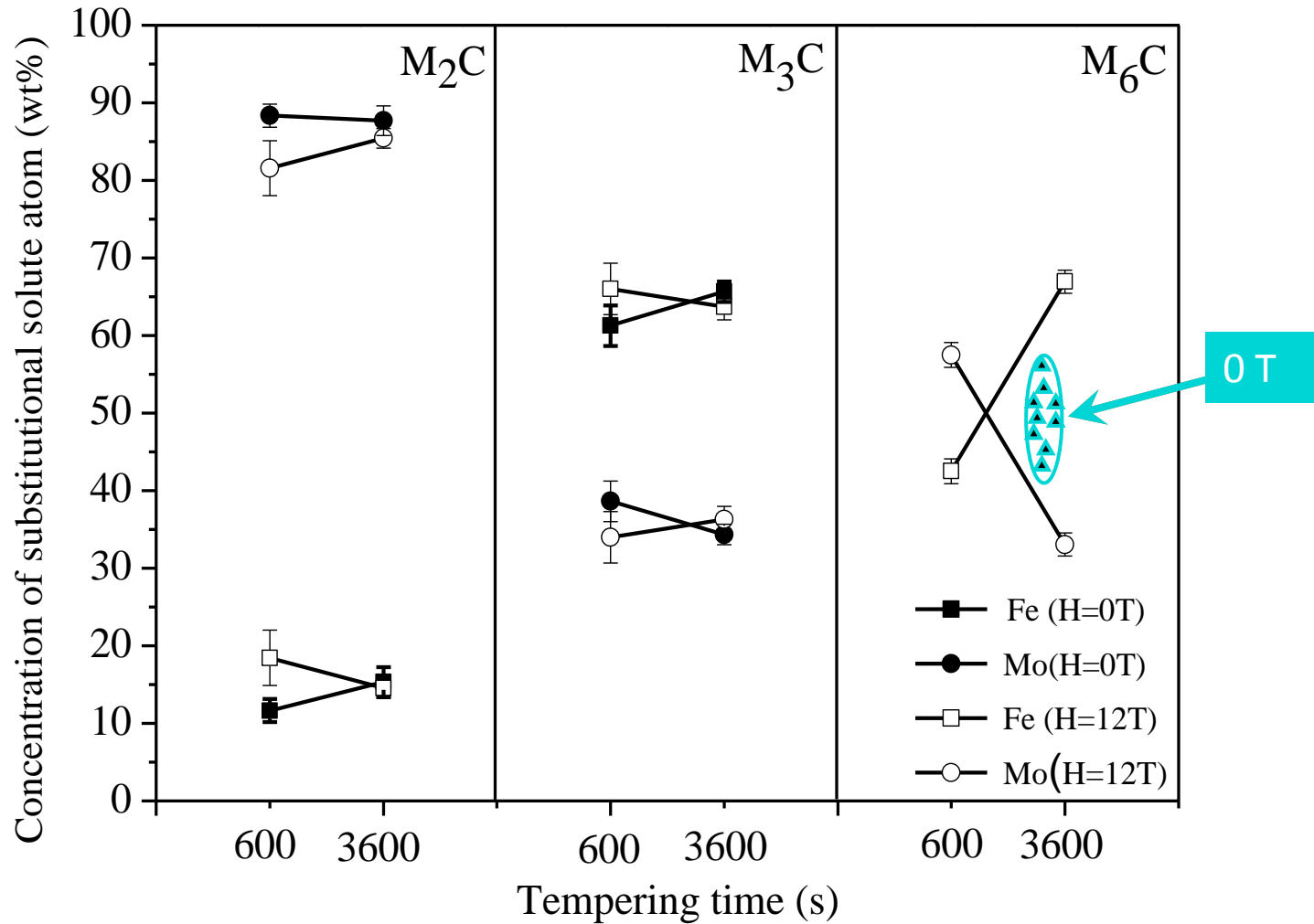
# Steel I : Fe-C-Mo, Tempered at 530 °C



	0 T		12 T	
Time	600 s	3600 s	600 s	3600 s
Type	$M_2C$ , $M_3C$	$M_2C$ , $M_3C$	$M_2C$ , $M_3C$ , $M_6C$	$M_2C$ , $M_3C$ , $M_6C$

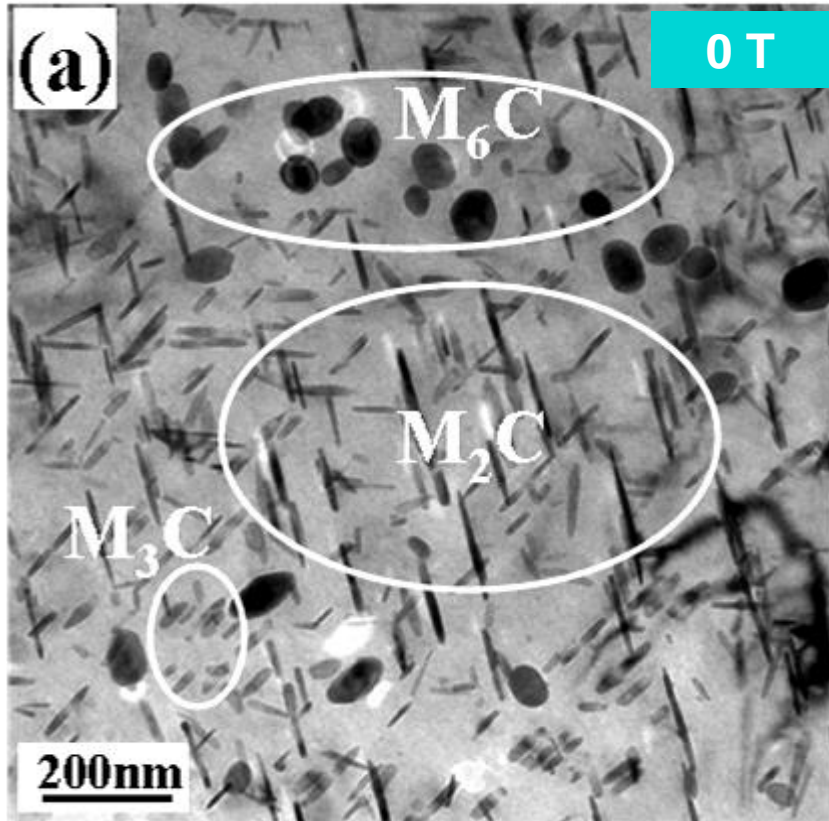
High magnetic field promotes the precipitation of  $M_6C$

# Steel I : Fe-C-Mo, Tempered at 530 °C

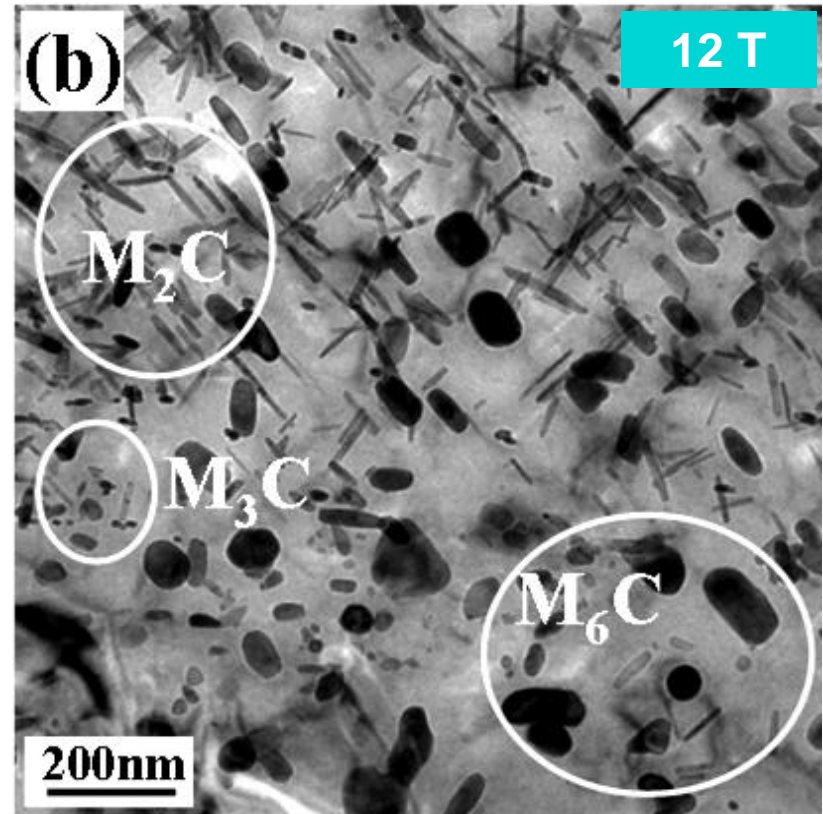


For  $M_2C$  and  $M_3C$ , no obvious change in concentration  
For  $M_6C$ , the Fe atom concentration increases

# Steel I : Fe-C-Mo, Tempered at 700 °C



700°C, 3600 s



700°C, 3600 s

High magnetic field has no visible influence on alloy carbide precipitation during high temperature tempering

# Steel II : 2.25Cr-Mo, Tempered at 550 °C

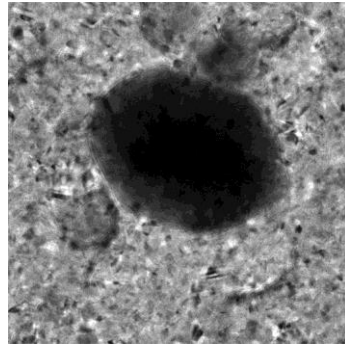
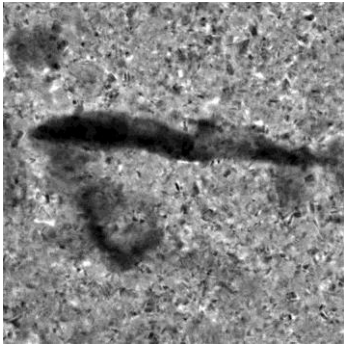
Tempered at 550°C for 600 and 3600 s

Without a magnetic field			With a 12-T magnetic field	
Time	600 s	3600 s	600 s	3600 s
Type	M <sub>2</sub> C, M <sub>3</sub> C	M <sub>2</sub> C, M <sub>3</sub> C	M <sub>2</sub> C, M <sub>3</sub> C M <sub>7</sub> C <sub>3</sub> , M <sub>23</sub> C <sub>6</sub>	M <sub>2</sub> C, M <sub>3</sub> C M <sub>7</sub> C <sub>3</sub> , M <sub>23</sub> C <sub>6</sub>
Carbides	(Fe, Cr, Mo) <sub>2</sub> C (Fe, Cr, Mo) <sub>3</sub> C	(Fe, Cr, Mo) <sub>2</sub> C (Fe, Cr, Mo) <sub>3</sub> C	(Fe, Cr, Mo) <sub>2</sub> C (Fe, Cr, Mo) <sub>3</sub> C (Fe, Cr, Mo) <sub>7</sub> C <sub>3</sub> (Fe, Cr, Mo) <sub>23</sub> C <sub>6</sub>	(Fe, Cr, Mo) <sub>2</sub> C (Fe, Cr, Mo) <sub>3</sub> C (Fe, Cr, Mo) <sub>7</sub> C <sub>3</sub> (Fe, Cr, Mo) <sub>23</sub> C <sub>6</sub>

High magnetic field promotes the precipitation of M<sub>23</sub>C<sub>6</sub> and M<sub>7</sub>C<sub>3</sub> carbides

# Steel II : 2.25Cr-Mo, Tempered at 550 °C

Gold replicas



$$N_{//} = \frac{4\pi}{k^2 - 1} \left[ \frac{k}{\sqrt{k^2 - 1}} \ln(k + \sqrt{k^2 - 1}) - 1 \right]$$

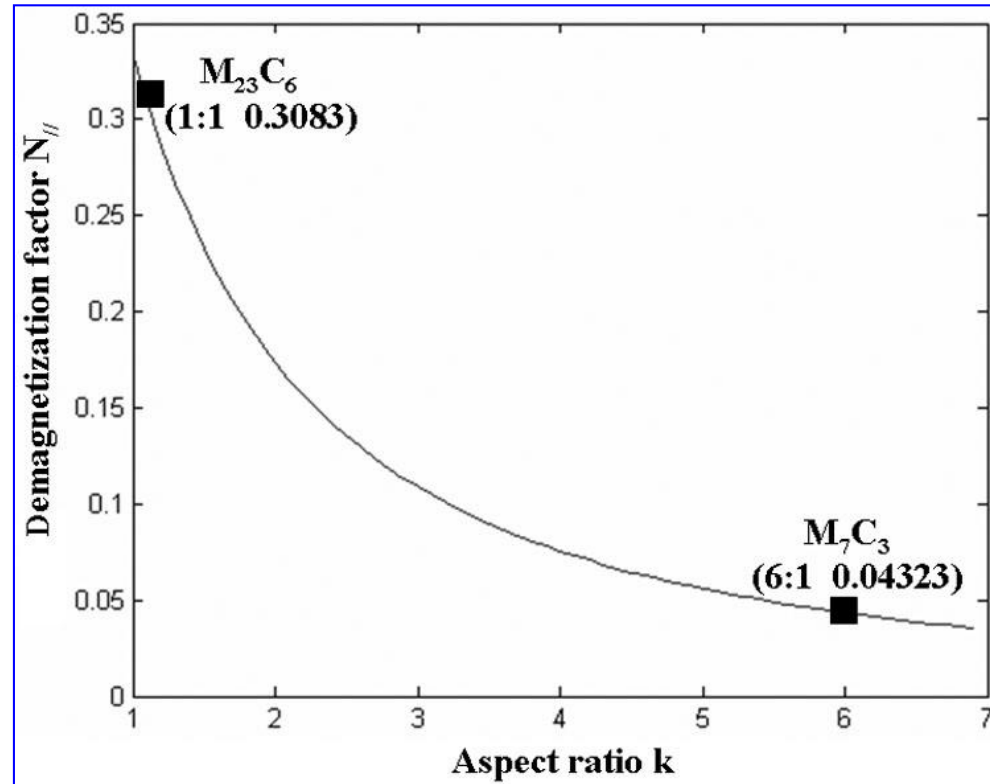
$M_7C_3$  6:1

$M_{23}C_6$  1:1

$(Cr_{2.24}Fe_{4.76})C_3$

$(Cr_{4.02}Fe_{18.98})C_6$

$M_{23}C_6$	Fe	Cr	Mo	Mn
Thermo-Cal	25.2	49.8	19.5	0.03
J-MatPro	25.1	48.3	19.4	0.25
Experimental	70.35	14.16	9.30	6.19



The increased Fe content in the  $M_{23}C_6$  and  $M_7C_3$  carbide

No visible influence on alloy carbide morphology

# Steel II : 2.25Cr-Mo, Tempered at 200, 700 °C

	0 T		12 T	
Time	600 s	3600 s	600 s	3600 s
Type	$M_2C, M_3C, M_7C_3$	$M_2C, M_3C, M_7C_3$	$M_{23}C_6$	$M_{23}C_6$

**200°C: The magnetic field promotes the precipitation of  $M_{23}C_6$**

	0 T, 12 T
Time	600 s, 3600 s
Type	$M_2C, M_3C, M_7C_3, M_{23}C_6$

**700°C: Alloy carbides are changed into paramagnetic state. High magnetic field has no visible influence on alloy carbide precipitation.**



$$G^*(T, H) = G^*(T, 0) - \int_0^M B \cdot dM$$

Chemical Gibbs free energy dependent on temperature

Magnetic Gibbs free energy mainly dependent on magnetic field strength and magnetization

# Discussion: Weiss molecular field theory

$$B = B_0 + \lambda M$$

$$M = NmB_j(\alpha(T))$$

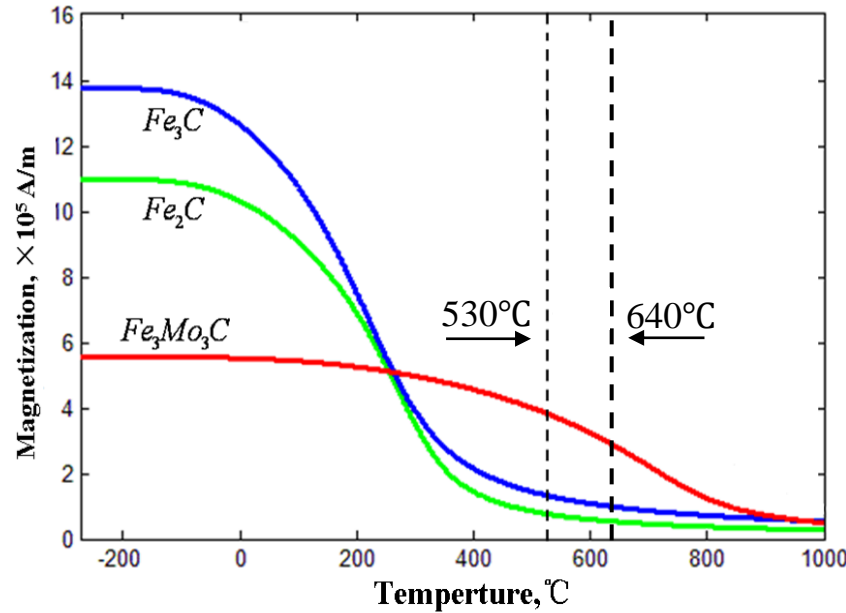
$$B_j(\alpha) = \left\{ \frac{2j+1}{2j} \operatorname{cth} \frac{(2j+1)a_j}{2j} - \frac{1}{2j} \operatorname{cth} \left( \frac{a_j}{2j} \right) \right\}$$

$$\alpha_j = \frac{n_B \mu_B B}{kT}$$

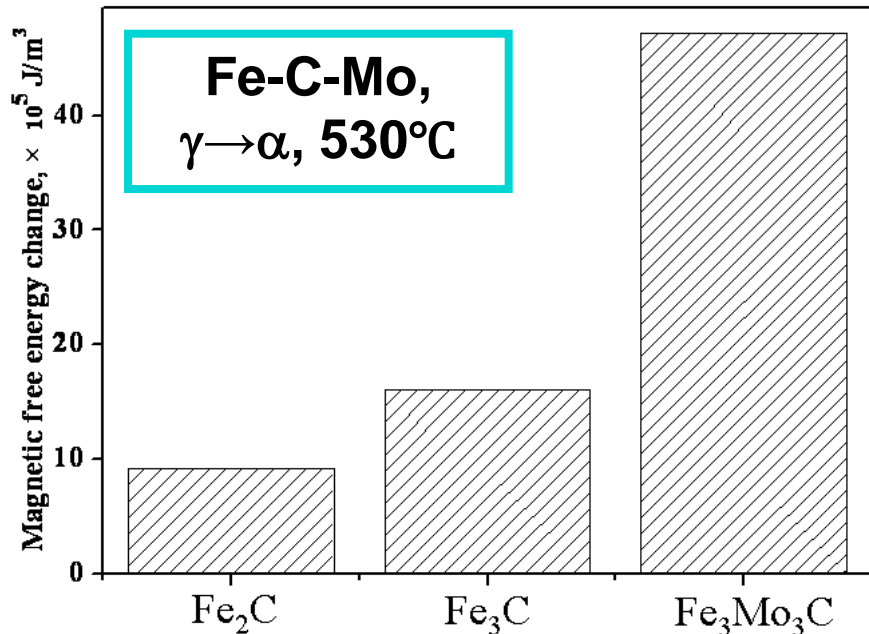
$$T_c = \frac{(j+1)Nn_B^2\mu_B^2\lambda}{3jk}$$

The theoretical calculation of the magnetization (M) with temperature (T)

# Magnetic field promotes the $M_6C$ precipitation

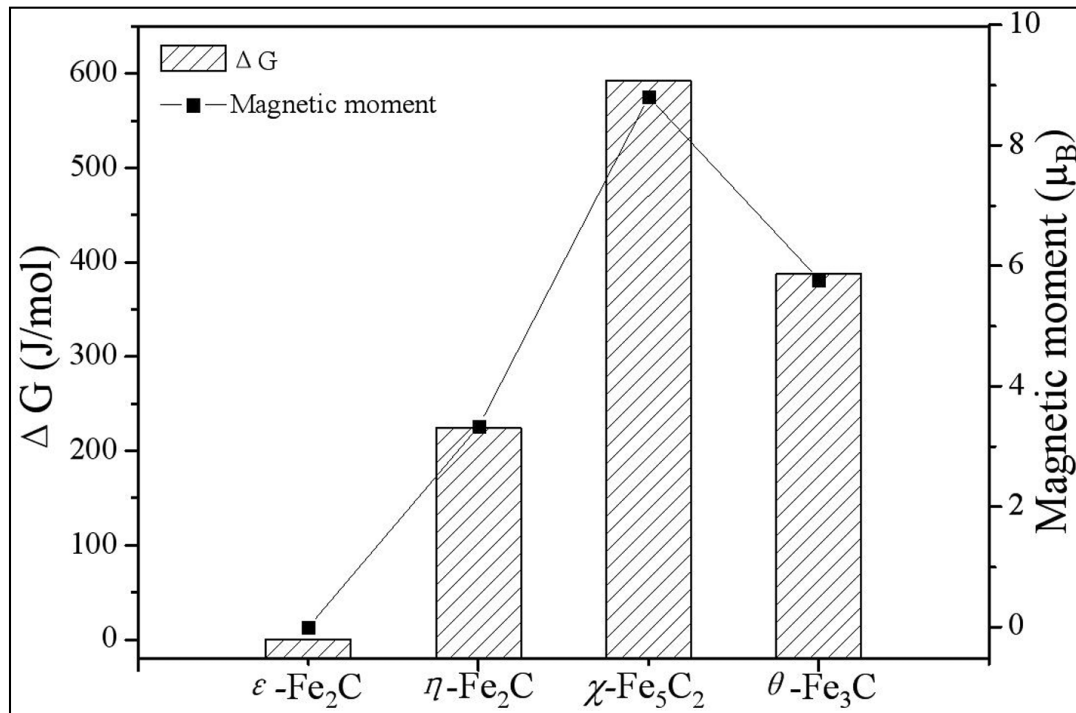


M-T curves



$$\Delta G^M = - \int_0^M \vec{B} \cdot d\vec{M}$$

# Magnetic field promotes the $\text{Fe}_5\text{C}_2$ Precipitation

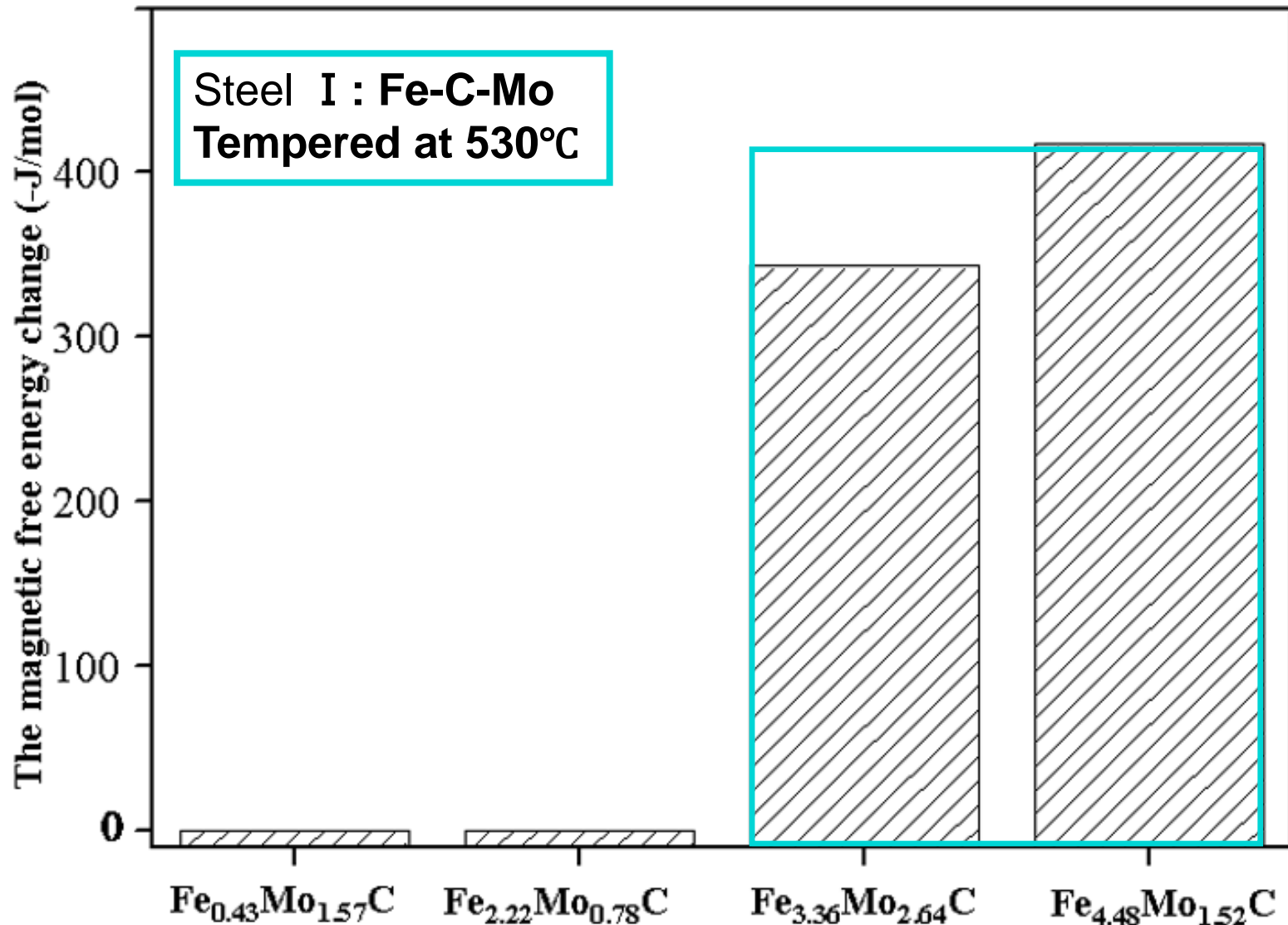


The magnetic free energy change and the magnetic moments

The magnetic free energy change of  $\chi\text{-Fe}_5\text{C}_2$  is the most remarkable

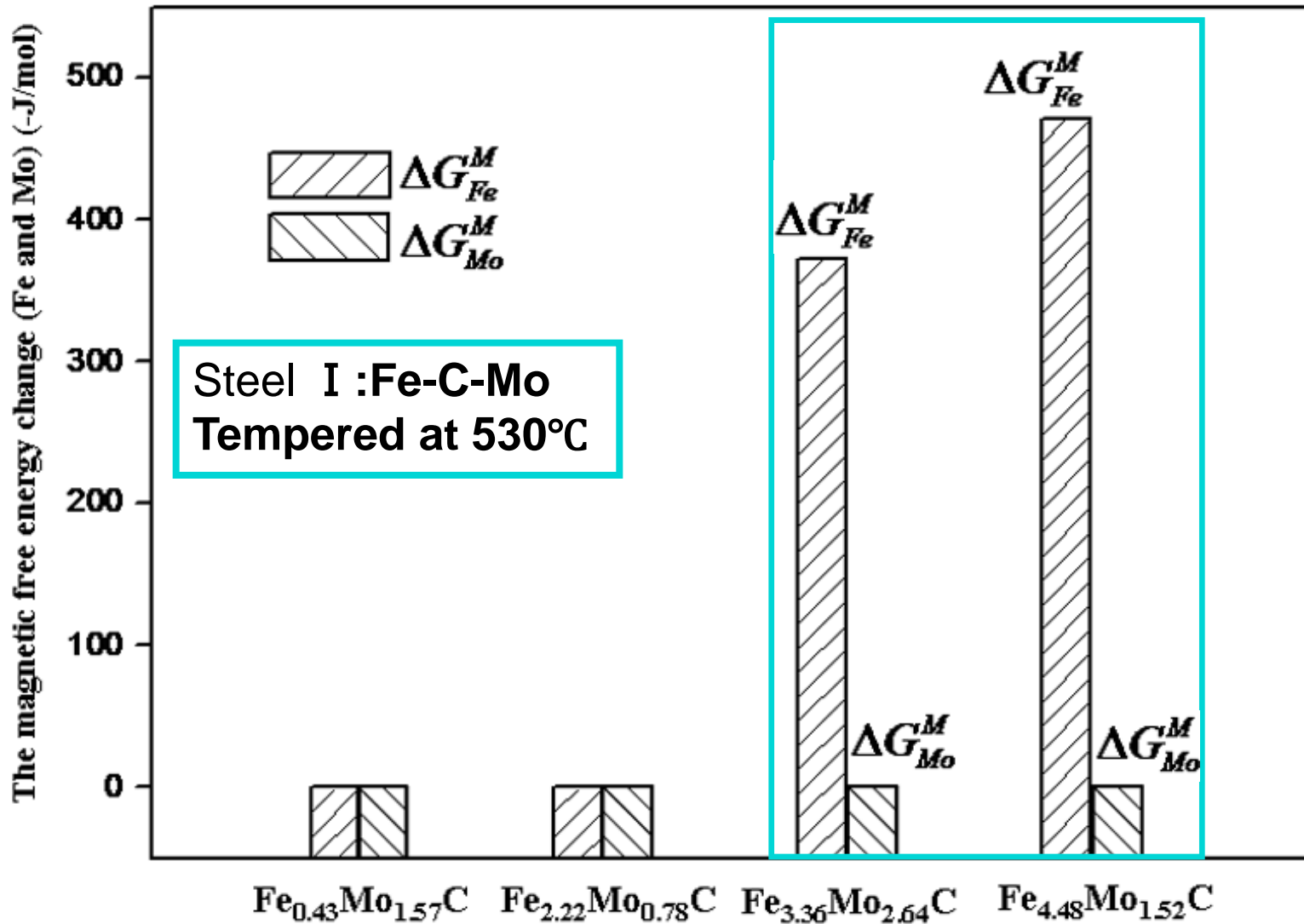
Steel I : Fe-C-Mo, Tempered at 200°C

# Magnetic field promotes the $M_6C$ precipitation



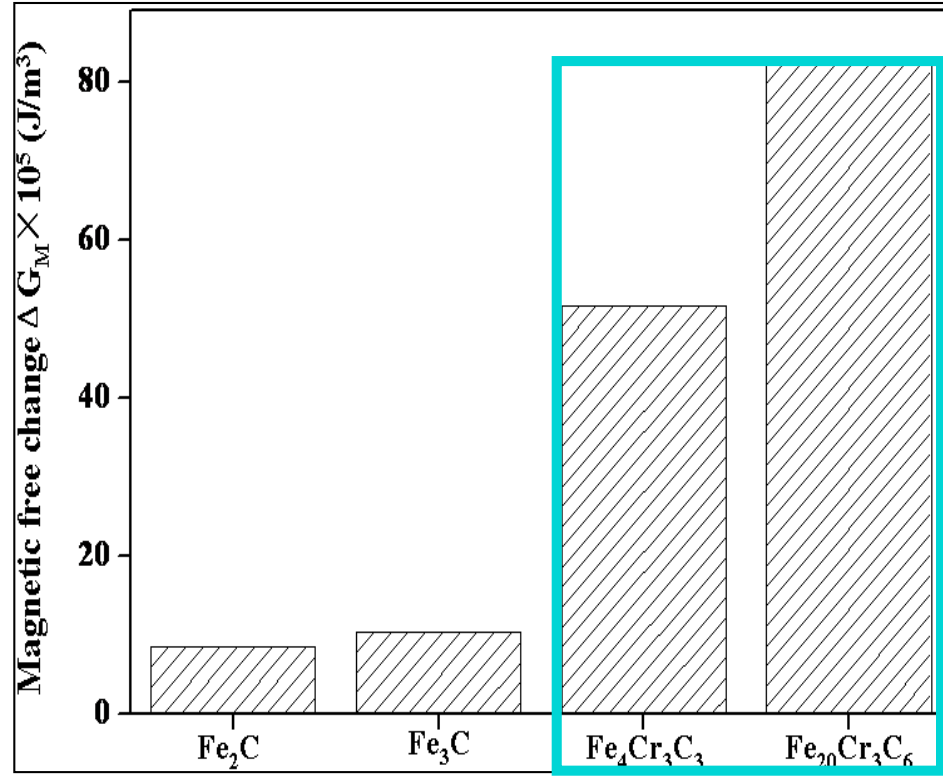
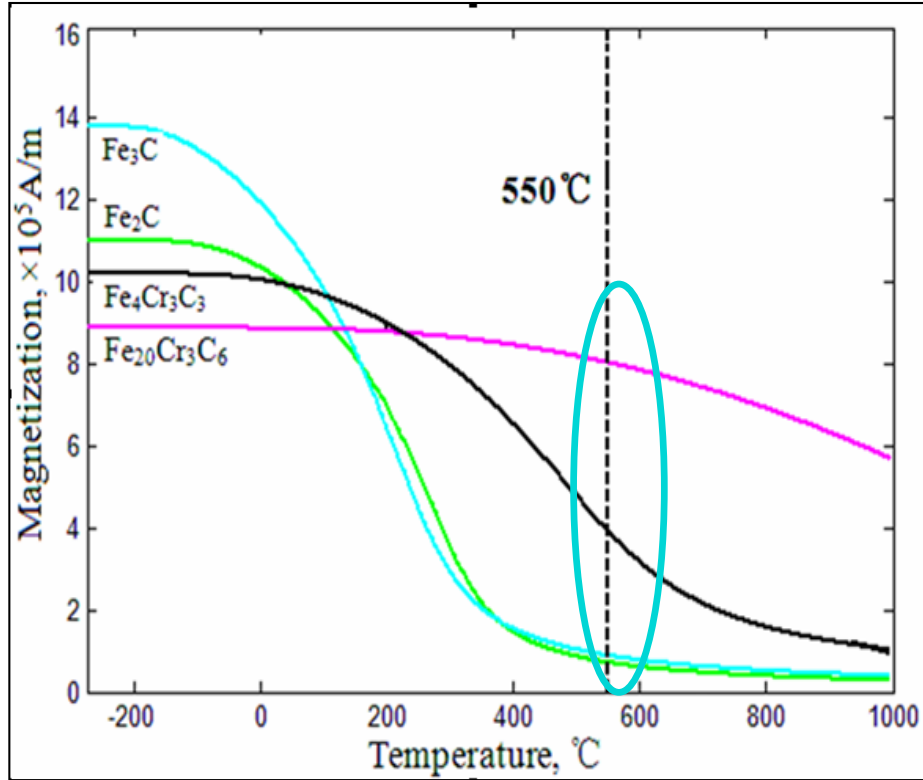
The magnetic free energy change of  $(Fe,Mo)_6C$  is the most remarkable

# Magnetic field increases the Fe concentration



The magnetic free energy change of Fe is increased remarkably

# M-T curves of $M_{23}C_6$ and $M_7C_3$ , and their magnetic Gibbs free energy



The magnetization curve with the temperature

The magnetic free energy change

High magnetic field promotes the precipitation of  $M_{23}C_6$  and  $M_7C_3$

Steel II : 2.25Cr-Mo, Tempered at  $550^{\circ}$ C

# Conclusions

The effect of high magnetic field on alloy carbide precipitation behaviors:

- The precipitation sequence of specific alloy carbide is changed
- The content of substitutional solute atom of Fe is increased
- No visible influence on alloy carbide morphology

The above three aspects are attributed to the magnetic free energy change with the presence of high magnetic field.